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CheKine[™] Micro Water and Soil Nitrite Content Assay Kit

Cat #: KTB3050

Size: 96 T

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REF	Cat #: KTB3050	LOT	Lot #: Refer to product label
	Applicable samples: Water, Soil		
X	Storage: Stored at 4°C for 12 months, protected from light		

Assay Principle

Nitrite exists widely in water and soil, which is not only an important intermediate of organic nitrogen decomposition, but also may come from pollution. Excessive intake of the human body can induce canceration of the digestive system. CheKine[™] Micro Water and Soil Nitrite Content Assay Kit provides a simple, convenient and rapid method for the determination of nitrite content, which is suitable for water and soil samples. The detection principle is that under acidic conditions, nitrite reacts with p-aminobenzene sulfonic acid to form diazo compounds, and then reacts with Nmuri 1-naphthyl ethylenediamine to form purplish red azo compounds with a characteristic absorption peak at 540 nm.

Materials Supplied and Storage Conditions

Kit components	Size	Storage conditions
Reagent	100 mL	4°C
Reagent II	10 mL	4°C, protected from light
ReagentIII	10 mL	4°C, protected from light
NaNO ₂ Standard	500 µL	4°C

Materials Required but Not Supplied

- Microplate reader or visible spectrophotometer capable of measuring absorbance at 540 nm
- 96-well plate or microglass cuvette, precision pipettes, disposable pipette tips
- · Ice maker, centrifuge, 2 mm sieve mesh
- Deionized water

Reagent Preparation

Reagent I : Ready to use as supplied. Equilibrate to room temperature before use. Store at 4°C.

Reagent II: Ready to use as supplied. Equilibrate to room temperature before use. Store at 4°C, protected from light.



Reagent III: Ready to use as supplied. Equilibrate to room temperature before use. Store at 4°C, protected from light. **Standard Curve Setting:** 1 mmol/mL Standard was diluted to 10 µmol/mL with Reagent I. Dilute the 10 µmol/mL Standard with Reagent I to 5, 2.5, 1.25, 0.625, 0.313, 0.156 µmoL/mL as indicated in the table below.

Num.	Volume of Standard	Volume of Reagent ι (μL)	The Concentration of Standard (μmol/mL)
Std.1	200 µL of Standard	0	10
Std.2	100 μL of Std.1 (10 μmol/mL)	100	5
Std.3	100 μL of Std.2 (5 μmol/mL)	100	2.5
Std.4	100 μL of Std.3 (2.5 μmol/mL)	100	1.25
Std.5	100 μL of Std.4 (1.25 μmol/mL)	100	0.625
Std.6	100 μL of Std.5 (0.625 μmol/mL)	100	0.313
Std.7	100 μL of Std.6 (0.313 μmol/mL)	100	0.156

Sample Preparation

1. Soil samples: Take appropriate amount of soil samples, remove stones, branches and other impurities, filter with 2mm sieve mesh, accurately weigh 0.5 g, add 1 mL Reagent |, shake at room temperature for 1 h, 8,000g, centrifuge 15 min at 25°C, and take the supernatant to be tested.

2. Water sample: Direct detection. If it is turbid, it can be determined after centrifugation.

Assay Procedure

1. Preheat the microplate reader or visible spectrophotometer for more than 30 min, and adjust the wavelength to 540 nm, visible spectrophotometer was returned to zero with deionized water.

2. Sample measurement. (The following operations are operated in the 96-well plate or microglass cuvette)

Reagent	Blank Well (µL)	Standard Well (μL)	Test Well (μL)
Reagent I	70	0	0
Standard	0	70	0
Sample	0	0	70
Reagent II	65	65	65
ReagentIII	65	65	65

3. Mix well and then incubatefor 15 min at 25°C. Then reading the values at 540 nm. The absorbance of blank well, standard well, test well recorded as A_{Blank} , $A_{Standard}$ and A_{Test} . Finally, calculate ΔA_{Test} - A_{Blank} , $\Delta A_{Standard}$ - A_{Blank} .

Note: Blank well and standard well only need to measure 1 time. In order to guarantee the accuracy of experimental results, need to do a pre-experiment with 2-3 samples. If ΔA_{Test} is less than 0.001, increase the sample quantity appropriately. If ΔA_{Test} is greater than 2.0, the sample can be appropriately diluted with Reagent I, the calculated result multiplied by the dilution factor, or decrease the sample quantity appropriately.

Data Analysis

Note: We provide you with calculation formulae, including the derivation process and final formula. The two are exactly equal. It is suggested that the concise calculation formula in bold is final formula.

1. Drawing of standard curve



With the concentration of the Standard solution as the y-axis and the $\Delta A_{Standard}$ as the x-axis, draw the standard curve. Substitute

the ΔA_{Test} into the equation to obtain the y value (µmoL/mL).

2. Calculate the content of NO²⁻ in sample

(1) Soil samples:

 $NO^{2-} (\mu mol/g)=y \times V_{Sample} \div (W \times V_{Sample} \div V_{Total}) \times n=2y \times n$

(2) Water sample:

NO²⁻ (µmol/mL)=y÷V_{Total}×n**=y×n**

Where: V_{Sample}: Sample volume, 0.07 mL; W: sample weight, 0.5 g; V_{Total}: Add the volume of Reagent I, 1 mL; n: dilution factor.

Typical Data

Typical standard curve:



Figure1. Standard Curve for NO²⁻. Examples



Figure 2. NO²⁻ content in soil. Assays were performed following kit protocol.

Recommended Products

Catalog No.	Product Name
KTB3040	CheKine™ Micro Glutamate Synthase (GOGAT) Assay Kit
KTB3041	CheKine™ Micro Glutamic Acid Dehydrogenase (GDH) Assay Kit

Disclaimer

The reagent is only used in the field of scientific research, not suitable for clinical diagnosis or other purposes.

